

# Syntheses of procyanidin B2 and B3 gallate derivatives using equimolar condensation mediated by Yb(OTf)<sub>3</sub> and their antitumor activities

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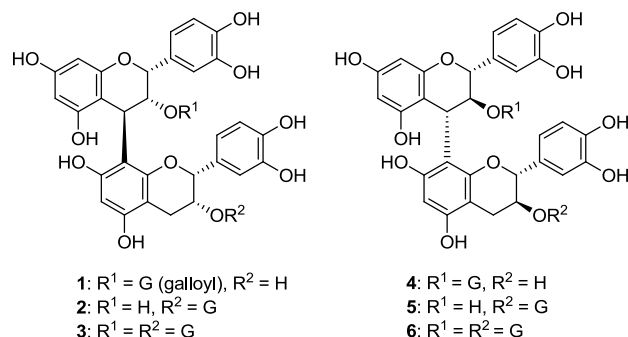
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**Abstract:** Synthesis of procyanidin B2 and B3 gallate derivatives, 3-*O*-gallate, 3''-*O*-gallate, and 3,3''-di-*O*-gallate, were synthesized using equimolar condensation mediated by Yb(OTf)<sub>3</sub>. Synthesized compounds showed significant antitumor effects against human prostate PC-3 cell lines. Their activities were weaker than well-known EGCG and prodelphinidin B3.

**Key words:** polyphenol, condensation, Lewis acid, procyanidin gallate, antitumor

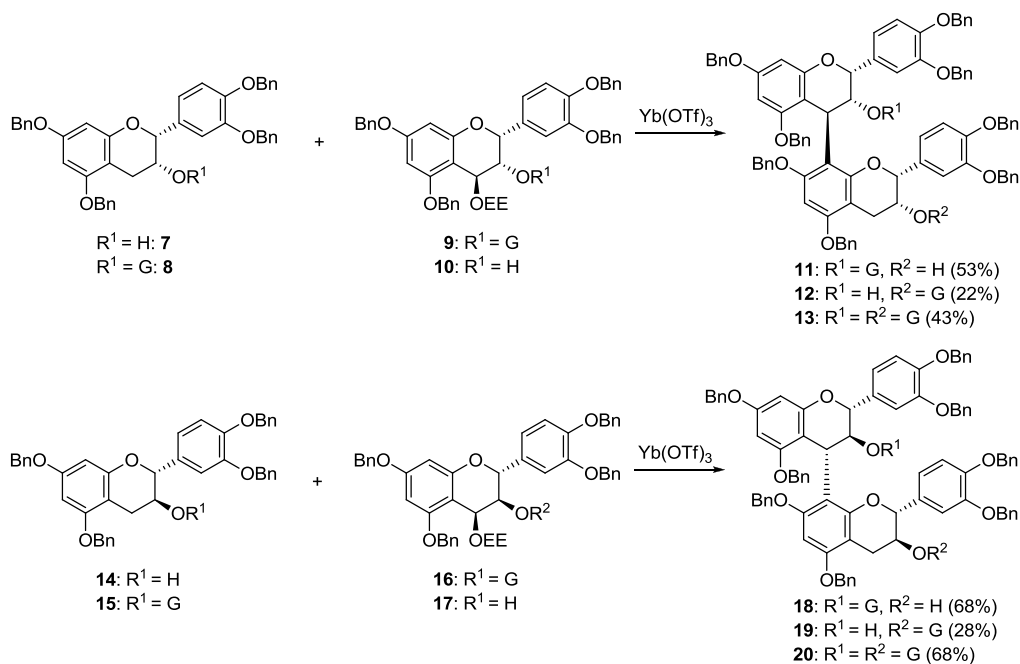
Procyanidin gallates are paid attention due to their significant biological activities.<sup>1</sup> For example, inhibitory activity of DNA polymerase<sup>2-4</sup> and antitumor activity were reported.<sup>5</sup> As to the procyanidin B2 and B3 gallates, a number of isolation and biological studies have been reported; procyanidin B2 3-*O*-gallate (**1**),<sup>5-10</sup> B2 3''-*O*-gallate (**2**),<sup>6-9,11-17</sup> B2 3,3''-*O*-di-gallate (**3**),<sup>12,13</sup> and B3 3-*O*-gallate (**4**).<sup>18,19</sup> Furthermore, many procyanidin gallates including various oligomer have been reported. However, it is difficult to isolate procyanidin gallates in pure state from the nature, the structural-activity relationship study (SAR study) of these compounds have not been clarified yet. Thus, synthetic studies on procyanidin gallates are very important to obtain them in pure state for the biological study. By far, the report of synthesis of procyanidin gallates are quite limited.<sup>2-4,20,21</sup> The reported syntheses of procyanidin gallates was accomplished using Lewis acid-mediated condensation of catechin and/or epicatechin derived nucleophiles and electrophiles. The disadvantage of this reaction is that using an excess

amount of nucleophile was necessary to avoid polymerization. So far, we have developed an efficient synthesis of procyanidin dimers through equimolar condensation of catechin nucleophiles with electrophiles using  $\text{Yb}(\text{OTf})_3$  as a Lewis acid.<sup>22-25</sup> Herein we demonstrate equimolar condensation of a catechin and/or a epicatechin nucleophile containing gallate moiety with a catechin and/or epicatechin derived electrophile and synthesis of procyanidin B2 and B3 gallates (Figure 1).



**Figure 1.** The structures of procyanidin B3 and B2 gallates (1-6).

As to the synthesis of procyanidin B2 gallates (1-3), the nucleophile **7**, **8** and the electrophile **9**, **10** were prepared by the reported procedure.<sup>2-4,22,23</sup> Equimolar condensation between nucleophile **7** and electrophile **9** using  $\text{Yb}(\text{OTf})_3$  as a Lewis acid to afford **11** in 53% yield. The benzylated procyanidin B2 3''-*O*-gallate (**12**) and procyanidin B2 3,3''-*O*-gallate (**13**) were also obtained from the condensation between nucleophile **8** and electrophile **10**, nucleophile **8** and electrophile **9** in 22% and 43% yields, respectively. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR data of the all condensed products **11-13** were identical with the reported values.<sup>3</sup>

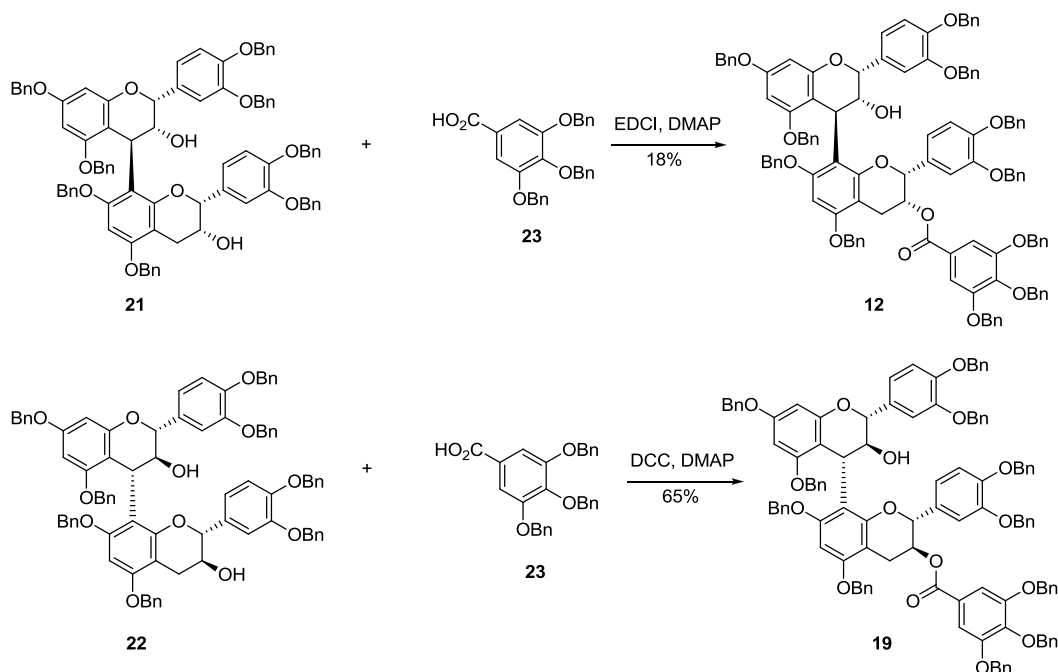


**Scheme 1.** Synthesis of benzylated procyanidin B2 and B3 gallates via  $\text{Yb}(\text{OTf})_3$  mediated equimolar condensation.

Similarly, related compounds of the procyanidin B3 gallates were synthesized. The nucleophile **14**, **15** and

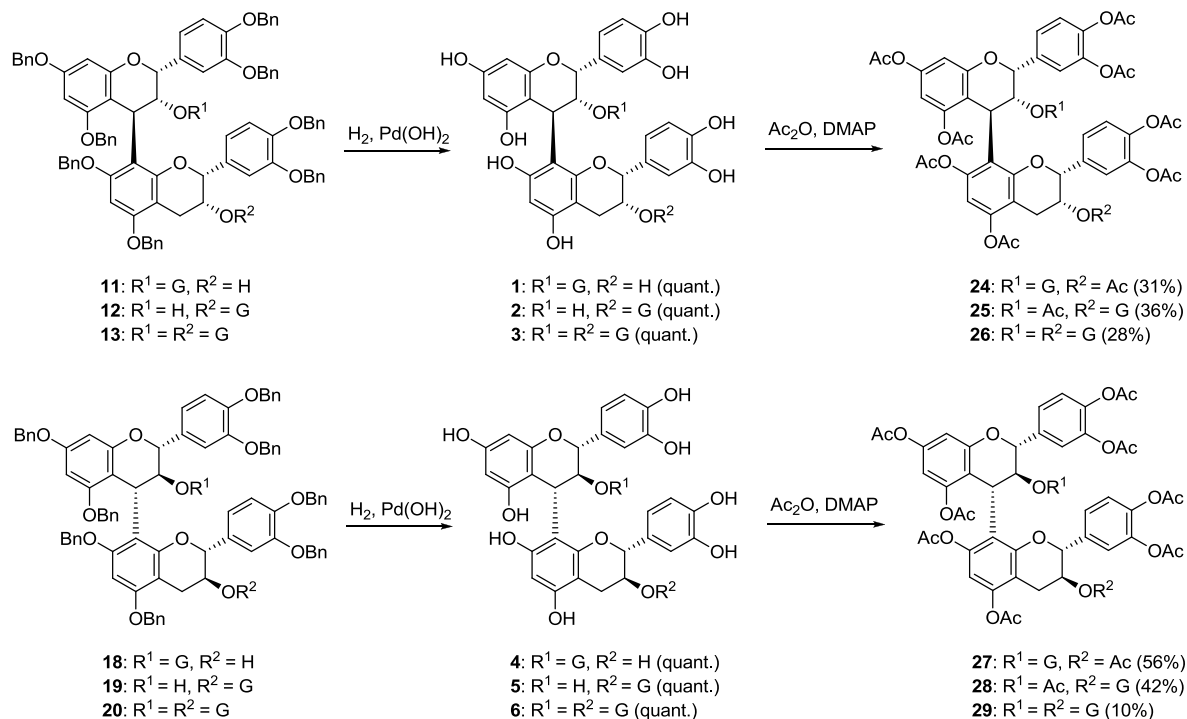
the electrophile **16**, **17** were prepared by the reported procedure.<sup>2-4,22,23</sup> Equimolar condensation between nucleophile **14** and electrophile **16** using Yb(OTf)<sub>3</sub> as a Lewis acid to afford **18** in 68% yield. The benzylated procyanidin B2 3''-O-gallate (**19**) and procyanidin B2 3,3''-O-di-gallate (**20**) were also obtained from the condensation between nucleophile **15** and electrophile **17**, nucleophile **15** and electrophile **16** in 28% and 68% yields, respectively. The <sup>1</sup>H and <sup>13</sup>C NMR data of the all condensed products **18-20** were identical with the reported values (Scheme 1).<sup>2,4</sup>

Because the yields of condensation between **8** and **10** (22%), **15** and **17** (28%) were low, we investigated the esterification of benzylated procyanidin B2 (**21**) and B3 (**22**) with benzylated gallic acid (**23**). In the case of the esterification between **21** and **23** using EDCI, the yield of desired product **12** was only 18% along with **11** (25%) and **13** (15%). On the other hand, esterification between **22** and **23** using DCC gave **19** in 65% yield. This method improved the synthesis of **19** compared to Yb(OTf)<sub>3</sub> mediated equimolar coupling (Scheme 2).



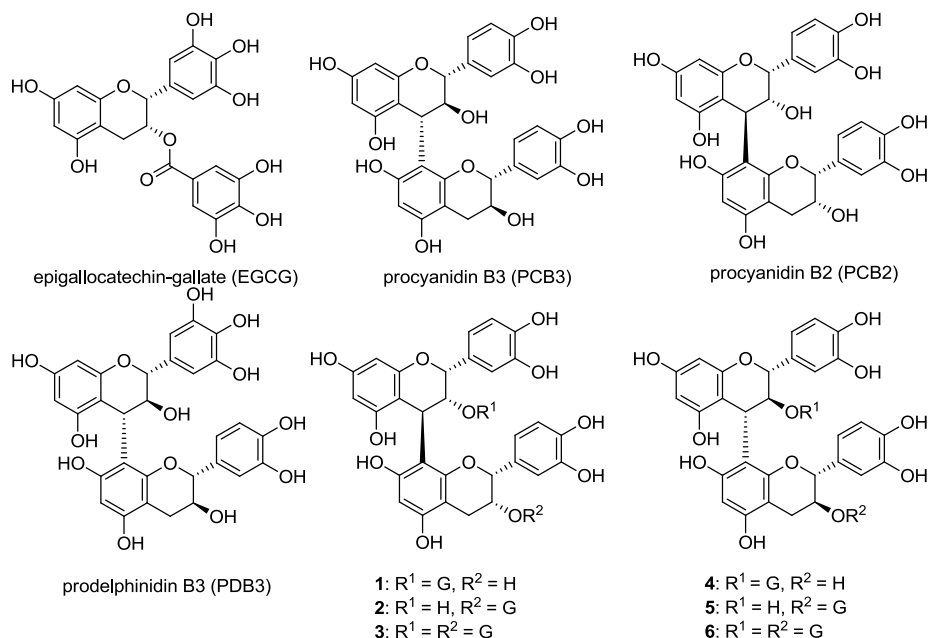
**Scheme 2.** Esterification of benzylated procyanidin B2 (**21**) and B3 (**22**) with benzylated gallic acid (**23**).

Deprotection of the benzyl ethers of **11-13** and **18-20** and subsequent lyophilization afforded **1-6** in good yield. We confirmed that lyophilized **1-6** was pure by HPLC analysis.<sup>26</sup> The specific rotation value of synthetic **1-6** were similar to those reported values.<sup>2,3</sup> We found that low concentration was necessary for measurements because of low degree of solubility when acetone was used as a solvent. Thus, we supplied the data of optical rotation values using MeOH as a solvent in the supplementary data. The <sup>1</sup>H and <sup>13</sup>C NMR data were reported in the literature using acetone-*d*<sub>6</sub>.<sup>11,18,19</sup> However, severe broadening of the peak was observed in the <sup>1</sup>H NMR spectra of **1-6** in our measurement. Thus we switched to use methanol-*d*<sub>4</sub> for measurement of NMR. As to the HRMS data of **1-6**, we obtained satisfied results. We transformed **1-6** into peracetate **24-29** to confirm the structures. The <sup>1</sup>H and <sup>13</sup>C NMR spectral data of peracetate **24-26** were in good agreement with the reported values.<sup>3</sup> Peracetate **27-29** also gave satisfied <sup>1</sup>H and <sup>13</sup>C NMR data (Scheme 3).



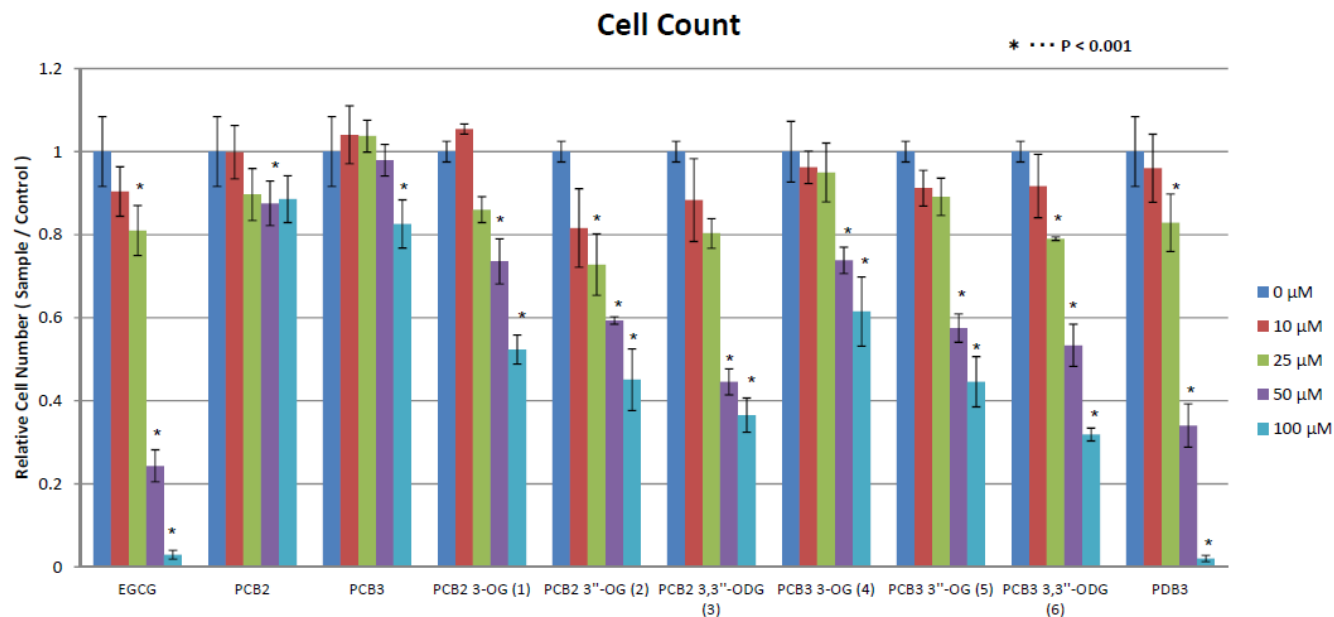
**Scheme 3.** Synthesis of **1-6** and their peracetate **24-29**.

Our interest was focused on examining the antitumor activities of the newly synthesized procyanidin B2 and B3 gallates. The synthesis of procyanidin B2 3-*O*-gallate (**1**), procyanidin B2 3''-*O*-gallate (**2**), procyanidin B2 3,3''-*O*-gallate (**3**), procyanidin B3 3-*O*-gallate (**4**), procyanidin B3 3''-*O*-gallate (**5**), and procyanidin B2 3,3''-*O*-gallate (**6**) allowed us to obtain sufficient quantities of purified compounds to screen against PC-3 prostate cancer cell lines together with procyanidin B2 (PCB2), B3 (PCB3) and prodelphinidin B3 (PDB3) which were prepared by us previously (Figure 2).<sup>22-25,27</sup>



**Figure 2.** The structures of test compounds for PC-3 prostate anticancer activity.

Results were obtained by cell count measurement. Epigallocatechin gallate (EGCG) was used as a positive control. As shown in Figure 3, procyanidin B2 3-*O*-gallate (PCB2 3-OG, **1**), procyanidin B2 3''-*O*-gallate (PCB2 3''-OG, **2**), procyanidin B2 3,3''-*O*-di-gallate (PCB2 3,3''-ODG, **3**), procyanidin B3 3-*O*-gallate (PCB3 3-OG, **4**), procyanidin B3 3''-*O*-gallate (PCB3 3''-OG, **5**), and procyanidin B2 3,3''-*O*-di-gallate (PCB3 3,3''-ODG, **6**) exhibited significant cytotoxic activity. The activities of procyanidin B2 3,3''-*O*-di-gallate (**3**) and procyanidin B2 3,3''-*O*-di-gallate (**6**) were stronger than those of mono-substituted gallate compounds. However, a comparison of the potencies of **3** and **6** with prodelphinidin B3 showed that the cytotoxic effects were a little bit weaker. This finding suggests that esterified pyrogallol moiety shows weaker activity compared to the compounds such as PDB3 (Figure 3).



After treatment of cells with PCB2 3-OG (**1**), PCB2 3''-OG (**2**), PCB2 3,3''-ODG (**3**), PCB3 3-OG (**4**), PCB3 3''-OG (**5**), and PCB3 3,3''-ODG (**6**) for 48 h, the cell proliferation was determined by cell count as shown in supplementary data. The values were represented as the rate of inhibition of cell proliferation by the treated sample compared to the untreated control (vehicle). Values are means  $\pm$  S.Ds. for three independent experiments. Asterisks indicated a significant difference between the control- and test-compound-treated cells, as analyzed by Student's test ( $p < 0.001$ ).

**Figure 3.** Effects of various concentrations of test compounds on cell proliferation.

In summary, syntheses of procyanidin B2 3-*O*-gallate (**1**), procyanidin B2 3''-*O*-gallate (**2**), procyanidin B2 3,3''-*O*-di-gallate (**3**), procyanidin B3 3-*O*-gallate (**4**), procyanidin B3 3''-*O*-gallate (**5**), and procyanidin B2 3,3''-*O*-di-gallate (**6**) were achieved via equimolar condensation of a catechin and/or an epicatechin nucleophile containing gallate moiety with a catechin and/or epicatechin derived electrophile. As to the synthesis of **5**, esterification between benzylated procyanidin B2 (**22**) and benzylated gallic acid (**23**) using DCC was effective. The antitumor activities of **1-6** suggested that these compounds showed significant cytotoxic activity but weaker than EGCG and prodelphinidin B3.

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## Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi>

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26. HPLC measurement condition: column; InertSustain C18 250×4.6 mm Waters, eluent 0.2% CH<sub>3</sub>COOH-CH<sub>3</sub>OH, flow rate: 0.8 mL/min, detection: UV 280 nm, retention time; procyanidin B2 3-*O*-gallate (**1**): 16.58 min, procyanidin B2 3''-*O*-gallate (**2**): 18.14 min, procyanidin B2 3,3''-*O*-gallate (**3**): 20.50 min., procyanidin B3 3-*O*-gallate (**4**): 17.87 min, procyanidin B3 3''-*O*-gallate (**5**): 25.18 min, procyanidin B2 3,3''-*O*-gallate (**6**): 24.86 min.
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