# Synthesis of Hypericin via Emodin Anthrone Derived from aTwo-fold Diels-Alder Reaction of 1,4-Benzoquinone 

Jiro Motoyoshiya*, Yusuke Masue, Yoshinori Nishi and Hiromu Aoyama<br>Department of Chemistry, Faculty of Textile Science and Technology, Shinshu University, Ueda, Nagano 386-8567, Japan<br>jmotoyo@giptc.shinshu-u.ac.jp

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#### Abstract

The six-step synthesis of hypericin by the regioselective two-fold Diels-Alder reaction of 1,4-benzoquinone first with (1-methoxy-3-methylbuta-1,3-dienyloxy)trimethylsilane leading to 7-methyljuglone, and next with (1,3-dimethoxybuta-1,3dienyloxy)trimethylsilane, to give emodin and its $O$-methylated derivative. The reduction of both compounds with tin(II) chloride in acidic media was accompanied by acid hydrolysis that produced emodin anthrone, whose oxidative dimerization with iron (III) chloride hydrate gave the bianthrone in high yield. The oxidation of the bianthrone in the presence of N ethyldiisopropylamine gave protohypericin, which was converted into hypericin upon irradiation.


Keywords: hypericin, emodin, emodin anthrone, Diels-Alder reaction.

Hypericin, 1, 3, 4, 6, 8, 13-hexahydroxy-10,11-dimethylphenanthro(1, 10, 9, 8-opqra)perylene-7, 14-dione (1), is isolated from St. John's-wort [1], a traditional medicinal plant distributed throughout the world, which has received much attention because of its various biological activities. Hypericin acts as an anticancer, antiviral, and antiretroviral agent, and induces apoptosis. Since its excited state generated by photo-irradiation produces active species such as singlet oxygen or superoxide ion that are toxic toward organisms, application in photodynamic therapy is considered to be a new use of hypericin [2]. This has prompted numerous reports on its synthesis [3]. However, the straightforward synthesis of hypericin from a simple compound has scarcely been documented because several steps are required to produce the key intermediates, emodin (2) [4a,b] or emodin anthrone (3) [5a]. The symmetric structure of $\mathbf{1}$ suggests that the dimerization of $\mathbf{2}$ or $\mathbf{3}$ by oxidative coupling is very convenient, and several syntheses are based on this synthetic strategy [5b-e].

In this paper, we report a short route to hypericin via emodin anthrone starting from 1,4-benzoquinone involving the two-fold Diels-Alder reaction. Our synthetic pathway to hypericin consists of a six-step synthetic procedure as shown in Scheme 1.



2


3

Either $\mathbf{2}$ or $\mathbf{3}$ is the key precursor of hypericin, and both are anthraquinone derivatives bearing a methyl group at the 6 -position and three hydroxyl groups at the 1,3 , and 8 -positions. The key problem of their synthesis is the introduction of such functionalities at the suitable positions. Calculations using the frontier molecular orbital theory predict that secondary orbital interaction controls the direction of cycloaddition [6a-c] from 1,4benzoquinone; namely, 7-methyljuglone formed by the initial Diels-Alder reaction of 1,4-benzoquinone and (1-methoxy-3-methylbuta-1,3-dienyloxy)trimethylsilane [6a,d] followed by acid hydrolysis is expected to react with (1,3-dimethoxybuta-1,3-dienyloxy)trimethylsilane [7] in a regioselective manner leading to the desired adduct, the emodin precursor.

The Diels-Alder reaction of 1,4-benzoquinone and (1-methoxy-3-methylbuta-1,3-dienyloxy)trimethylsilane was carried out in benzene at $50^{\circ} \mathrm{C}$ and the workup followed by acid hydrolysis afforded 7-methyljuglone

$446 \%$

$\xrightarrow[\mathrm{HCl} / \mathrm{CH}_{3} \mathrm{COOH}]{\mathrm{SnCl}_{2} \cdot \mathrm{HCl}} \quad 3 \quad 91 \%$



Scheme 1
(4) in 46 \% yield. The second Diels-Alder reaction of 4 and (1,3-dimethoxybuta-1,3-dienyloxy)trimethylsilane proceeded in benzene at $70^{\circ} \mathrm{C}$, and a mixture of physcion (5a) [4a] and emodin 6,8-dimethyl ether (5b) [4c] in $28 \%$ and $68 \%$, respectively, was afforded after a similar workup. Anthraquinones $\mathbf{5 a}$ and $\mathbf{5 b}$ are naturally occurring and were readily converted into 2 by acid hydrolysis. The structural elucidation of 5a and $\mathbf{5 b}$ was done by reduction with tin (II) chloride in acetic acid containing hydrochloric acid to 3 , whose NOE between the protons at the 10 -position, chemoselectively introduced by reduction of the carbonyl group, and the two peri-protons at the 4 -, and 5 -positions unambiguously established its structure. As expected, this second Diels-Alder reaction proceeded in a highly regioselective manner. The reduction was accompanied by acid hydrolysis leading to 3. Despite
the low yield of $\mathbf{4}$ by the first Diels-Alder reaction and sequential acid hydrolysis, the present synthetic method is not only simple, but also the most practical of those already reported for $2[3,8]$ and 3 [5a].

The conversion of $\mathbf{3}$ into $\mathbf{1}$ was completed by the previously established procedure [9] with modifications. The oxidative coupling of 3 using iron (III) chloride hydrate gave the bianthrone (6) [10] in $77 \%$ yield as a mixture of diastereomers, which was oxidized under an oxygen atmosphere in the presence of $N$-ethyldiisopropylamine to yield protohypericin (7) [5c,9] in $98 \%$ yield. This procedure was much improved by using the tertiary amine in an organic solvent. The final step was the photooxidation of 7 leading to $\mathbf{1}$. A solution of 7 in acetone was irradiated under a high-pressure mercury lamp to give $\mathbf{1}$ in $31 \%$ yield after purification by column chromatography. The spectral data are in agreement with those reported. The yield in this step was low, but not inferior to the procedure using 2 . The patent procedure [11] for the oxidation of protohypericin to hypericin leads to the improvement of the yield. Thus, the synthesis of hypericin was accomplished by a six-step synthetic procedure, starting from 1,4-benzoquinone.

## Experimental

General: Melting points were determined on a hot stage microscope apparatus (Mitamura). NMR spectra were recorded on a Bruker AVANCE-400 at 400 MHz and 100 MHz respectively. The chemical shifts ( $\delta$ ) are reported in ppm downfield from TMS as internal standard or from the residual solvent peak. Coupling constants ( $J$ ) are reported in Hz. Low resolution mass spectra (MS) were recorded by the JEOL JMS-K9 spectrometers. Elemental analysis was recorded on a Perkin-Elmer 2400CHN elemental analyzer. Analytical TLC was carried out on precoated silica gel 60F-254 plates (E. Merck). Column chromatography was performed on silica gel (E. Merck).

## 7-methyljuglone (4)

A solution of 1,4-benzoquinone ( $1.50 \mathrm{~g}, 13.4 \mathrm{mmol}$ ) was added to a solution of (1-methoxy-3-methylbuta-1,3-dienyoxy)trimethylsilane ( $1.26 \mathrm{~g}, 6.76 \mathrm{mmol}$ ) in benzene ( 8 ml ), and the combined solution was stirred at room temperature for 5 hr and then heated at $50^{\circ} \mathrm{C}$ overnight. After removal of the solvent, the residue was treated with aqueous hydrochloric acid (5\%, 32 ml ) in THF ( 10 ml ) at room temperature for 3 hr . The product obtained after removal of the organic solvent
was extracted with chloroform. The organic layer was dried over sodium sulfate and condensed in vacuo. The crude product was purified with silica gel column chromatography (chloroform) to afford 7methyljuglone ( $581 \mathrm{mg}, 46 \%$ ).

MP. $125-127^{\circ} \mathrm{C}$.
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 2.44$ (s, 3H, Me); 6.91 (s, 2H, H-2 and 3); 7.08 (brs, $1 \mathrm{H}, \mathrm{H}-6$ ); 7.44 (d, $1 \mathrm{H}, J=1.3 \mathrm{~Hz}, \mathrm{H}-$ 8); 11.87 (s, $1 \mathrm{H}, \mathrm{OH}) ;{ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 22.0\left(\mathrm{CH}_{3}\right)$, 112.8(CH), 120.3 (CH), 124.0 (CH), 131.3 (C) , 138.6 (C), 139.1 (C), 148.3 (CO), 161.5 (CO), 184.4;

Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{O}_{5}$ : C 70.21, H 4.29\%, Found : C 70.35, H 4.15\%.

Physcion (5a) and 6,8-0,0'-dimethylemodin (5b): A solution of $4(1.06 \mathrm{~g}, 5.62 \mathrm{mmol}$ ) in benzene ( 30 ml ) was added to (1,3-dimethoxybuta-1,3-dienyloxy)trimethylsilane ( $2.28 \mathrm{~g}, 11.2 \mathrm{mmol}$ ) in benzene ( 15 ml ) and the resulting solution was stirred at room temperature for 3 hr and then heated at $70^{\circ} \mathrm{C}$ for a day. The residue obtained after removal of the solvent was treated with aqueous hydrochloric acid $(5 \%, 32 \mathrm{~mL})$ in THF ( 45 ml ) at room temperature for a day. After removal of the organic solvent, the products were extracted with chloroform. The organic layer was dried over sodium sulfate and condensed in vacuo. The crude products were purified with silica gel column chromatography (chloroform) to give $\mathbf{5 a}(0.45 \mathrm{~g}, 28 \%)$ and $5 \mathbf{5}(1.14 \mathrm{~g}, 68 \%)$.

## Physcion

MP: $203^{\circ} \mathrm{C}$.
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): 2.45$ (s, 3H, Me), 3.94 (s, $3 \mathrm{H}, \mathrm{OMe}$ ), 6.69 (d, 1H, J = 2.52, H-2), 7.09 (s, 1H, H-7), 7.37 (d, $1 \mathrm{H}, \mathrm{J}=2.78, \mathrm{H}-4$ ), 7.63 (s, 1H, H-5), 12.11 (s, 1H, $\mathrm{OH}), 12.31$ (s, 1H, OH);
${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ : $22.5\left(\mathrm{CH}_{3}\right), 56.4\left(\mathrm{OCH}_{3}\right), 107.1$, 108.5, 114.0, 121.6, 122.4, 124.8, 133.5, 135.6, 148.8 (C), 162.8(C), 165.5 (C), 166.9 (C), 182.2 (CO), 191.1 (CO). Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{O}_{5}$ : C 67.70, H $4.25 \%$. Found: C 67.74, H 4.16\%.

## 6,8-O,O'-dimethylemodin

MP: 211-213 ${ }^{\circ} \mathrm{C}$ (Lit. 213-214 ${ }^{\circ} \mathrm{C}$ [4c]).
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 2.43$ (s, 3H), 3.98 (s, 3H), 4.03 (s, 3H), 6.79 (d, 1H, J=2.5), 7.08 (s, 1H), 7.47 (d, 1H, $\mathrm{J}=2.5$ ), 7.57 (s, 1H), 13.08 (s, 1H);
${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 22.3,56.4,56.9,104.3,105.0$, 120.3, 123.0, 125.1, 132.7, 138.0, 147.2, 163.0, 163.3, 165.6, 183.3, 187.8. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{O}_{5}$ : C 68.45, H 4.73\%. Found: C 68.75, H 4.58\%.

Emodin anthrone (3) from 5a and 5b: A warm solution of tin (II) chloride hydrate ( $1.07 \mathrm{~g}, 4.78 \mathrm{mmol}$ ) in conc. hydrochloric acid ( 5.5 ml ) was added to a suspension of 5 ( $136 \mathrm{mg}, 0.478 \mathrm{mmol}$ ) in acetic acid ( 10 ml ) to give the red-colored solution. After the solution was heated at $120^{\circ} \mathrm{C}$ one day, it was poured into ice-water and the precipitate, 3 , was collected by filtration and dried ( $118 \mathrm{mg}, 96 \%$ ). The similar procedure using $5 \mathbf{5}$ ( $231 \mathrm{mg}, 0.774 \mathrm{mmol}$ ) in acetic acid ( 15 ml ), tin (II) chloride hydrate ( $1.07 \mathrm{~g}, 4.78$ mmol ) in conc. hydrochloric acid ( 9.2 ml ) gave 3 (189 $\mathrm{mg}, 95 \%$ ).

MP: 254-258 ${ }^{\circ} \mathrm{C}$ (Lit. $254-258^{\circ} \mathrm{C}$ [5a]).
${ }^{1} \mathrm{H}-\mathrm{NMR}$ (DMSO- $\mathrm{d}_{6}$ ): $\delta_{\mathrm{H}}: 2.33$ (s, 3H, Me), 4.31 (s, 2H, CH2), 6.22 (d, 1H, $J=2.2 \mathrm{~Hz}, \mathrm{H}-4$ ), $6.43(\mathrm{~d}, 1 \mathrm{H}, J=$ 2.2, H-2), 6.69 (s, 1H, H-5), 6.79 (s, 1H, H-7), 10.80 (s, $1 \mathrm{H}, \mathrm{OH}$ ), 12.21 (s, 1H, OH), 12.37 (s, $1 \mathrm{H}, \mathrm{OH}$ );
${ }^{13} \mathrm{C}-\mathrm{NMR}$ (acetone- $\mathrm{d}_{6}$ ) 22.3 (CH3), 33.6(CH2), 102.4 (CH), 108.4 (CH), 110.1 (CH), 114.4 (CH), 116.6 (C), 121.0 (C),143.2 (C),146.3 (C), 148.5 (C), 163.9 (C),165.9 (C), 166.7 (C), 193.2 (CO).

Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{O}_{5}$ : C 70.31, H 4.72\%. Found: C 70.57, H 4.63\%.

Emodin (2) from 5a and 5b: After a solution of 5a ( $300 \mathrm{mg}, 1.05 \mathrm{mmol}$ ) and hydrobromic acid ( $48 \%, 59$ ml ) in acetic acid ( 59 ml ) was heated at $125^{\circ} \mathrm{C}$ for 12 h , ice-water was added and the precipitate was filtered. The dried product was purified with silica gel column chromatography (chloroform-ethyl acetate) to give 2 ( $258 \mathrm{mg}, 91 \%$ ). The similar procedure using $5 \mathbf{5 b}$ (300 $\mathrm{mg}, 1.01 \mathrm{mmol}$ ) and hydrobromic acid ( 29 mL ) in acetic acid ( 59 mL ) afforded 2 ( $129 \mathrm{mg}, 47 \%$ ) along with $\mathbf{5 a}(53 \mathrm{mg}, 18 \%)$.

MP: 256-258 ${ }^{\circ} \mathrm{C}$ (Lit. $257^{\circ} \mathrm{C}$ [4a]).
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) 2.45$ (s, 3H, Me), 5.74(s, 1H, OH), 6.67 (d, 1H, $J=2.2 \mathrm{~Hz}, \mathrm{H}-7$ ), 7.09 (d, 1H, $J=1.0 \mathrm{~Hz}$, $\mathrm{H}-5$ ), 7.30 (d, 1H, $J=2.5 \mathrm{~Hz}, \mathrm{H}-4$ ), 7.63 (d, 1H, $J=$ 1.7 Hz ), 12.09 (s, 1H, OH), 12.28 (s, 1H, OH).

Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{O}_{5}$ : C 66.67, H 3.73\%. Found: C 66.42, H 3.82\%.

Emodin bianthrone (6) [10] A solution of iron (III) chloride hydrate ( $127 \mathrm{mg}, 0.468 \mathrm{mmol}$ ) in ethanol ( 12 ml ) was added drop wise to a solution of emodin anthrone ( $100 \mathrm{mg}, 0.390 \mathrm{mmol}$ ) in ethanol ( 25 mL ) within 45 min . After heating at 4 h under reflux, the solution was poured into an aqueous hydrochloric acid $(5 \%, 50 \mathrm{ml})$ and the product was extracted with ether.

After being dried over sodium sulfate and removal of the solvent, the residue was purified with column chromatography (chloroform:methanol $=20: 1$ ) to give 6 (78 mg, 77\%) as a diastereomeric mixture.
${ }^{1} \mathrm{H}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): 2.27,2.31$ (s, 6H, Me), 4.35 (s, 2H, H-10, 10’), 5. 92, 6.02, 6.04, 6.13, 6.29, 6.33 (s, 4H, H2,2’, 5,5’, 7,7’), 6.71, 6.68 (s, 2H, H-4,4’), 11.78, 11.83, 12.12, 12.16 (s, 4H, OH).

Protohypericin (7) [5d] A solution of 6 (94 mg, 0.184 $\mathrm{mmol})$ and $N$-ethyldiisopropylamine $(0.12 \mathrm{ml}, 0.703$ mmol ) in acetonitrile ( 50 ml ) was bubbled with oxygen gas for 15 min and heated under reflux and an oxygen atmosphere for 2 h . After removal of the solvent, the solution was poured into aqueous hydrochloric acid $(10 \%, 65 \mathrm{ml})$ and the precipitate was washed with water. The collected precipitate was treated with acetone and the extract was concentrated in vacuo. The
product was purified with silica gel column chromatography (chloroform and acetone) to give 7 (146 mg, 98\%).
${ }^{1} \mathrm{H}-\mathrm{NMR}$ (acetone- $\mathrm{d}_{6}$ ): 2.11(s, 6H), 6.36 (s, 2H), 6.70 (s, 2H), 7.29 (s, 2H).

Hypericin (1) [2,5e] A solution of 7 (146 mg, 0.288 mmol) in acetone ( 20 ml ) was irradiated under a highpressure mercury lamp at room temperature for 1 h . After removal of the solvent, the residue was purified with silica gel column chromatography (chloroform : acetone $=1: 1$ ) to give 1 ( $45 \mathrm{mg}, 31 \%$ ).
${ }^{1} \mathrm{H}-\mathrm{NMR}$ (acetone- $\mathrm{d}_{6}$ ) $\delta: 2.79$ ( $\mathrm{s}, 6 \mathrm{H}, \mathrm{Me}$ ), 6.62, 7.36 (s, 4H, ArH); UV (EtOH, с $1.0 \times 10^{-5}$ ): l 591 (30940), 547 (15250), 475 (9370), 381 (9290), 329 (20540) nm ( $\varepsilon$ ).

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